NUCLEOSIDES, 45. - ASSIGNMENT OF GLYCOSYLATION SITES IN 0-HEXOPYRANOSYL-RIBONUCLEOSIDES BY 13C-NMR ')

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Summary: Evidence is presented that glycosylation of a ribose-OH group in nucleosides results
in a significant downfield shift for the appended ¹³C nucleus and smaller upfield displacements for the adjacent carbons, providing an efficient tool for differentiating between 2'-O-, 3'-O- and 5'-O-glycosyl-ribonucleoside. galactosylation of uridine, Therewith, the products formed on enzymatic inosine, and adenosine $4-6$) are unequivocally assigned $\beta(1+3)$ glycosidic linkages (3a - 3c).

The glycosidose-induced transfer of a hexose unit onto a ribose is marked by low regioselectivity yielding (l-3)- and (l--5)-linked hexosyl-riboses as major, the (l--2)- and (l&4)-coupled isomers as minor products²⁾. A higher selectivity appears to apply to ribonucleosides as hexosyl acceptors, since a-gluco**sidoses of either fungal, bacterial or plant origin transfer glucosyl moieties from maltose (or soluble starch)** to the ribose–5'–oxygen exclusively, resulting in glucosyl–nucleosides of type $\underline{\texttt{l}}^{(3)}$. E.coli–derived ß–galact **sidase, however, combined with o-nitrophenyl B-D-galactopyranoside as the hexosyl donor, glycosylotes odenosine4), inosine5), or uridine6) with a high (e.g. 10: 1 5'6)) preference for the secondary ribose** hydroxyls. The resulting minor components, i.e. 5'-O-galactosyl-nucleosides <u>2b, 2c</u> and <u>2d</u> have been structurally clarified by identification of hydrolysis products, by periodation studies as well as by unequi**vocal syntheses of \$J 7) and the underlying galactosyl-6 (l-5)-ribose 8)** ; **in contrast, the nature of the maior products, i.e. whether galoctosylation occured at the g-2' or o-3' of the ribose portion, remains to be unambiguously established, olthough some synthetic studies 9) and some subtile differences in 'H-NMR chemical shifts 9) strongly indicated them to be the galactopyranosyl- B (1+3)-ribonucleosides & - a. Unequivocal proof thereof is now provided by comparative evaluation of '3C-chemical shifts of the ribose** carbons in 3a and 3b with those of their parent nucleosides.

In $3a$ and $3b^{10}$, 13 C-NMR resonances for galactosyl and ribose carbons fall within the same range, yet the **former are readily identified by comparison with the signals obtained for methyl B-D-galactopyranoside (cf. Fig. 1), chemical shift differences for C-2 through C-5 of galactose being between 0.07 and 0.32 ppm only. Direct additional proof for the assignments in Fig. 1 is provided by selective decoupling of the** galactose protons ¹¹⁾, which, at 380 Hz (cf. Fig. 2), leaves the galactosyl carbons as discrete singlets. On **increasing the decoupling frequency, the ribose carbons successively become decoupled (Fig. 2), whereby the signal for C-3' turns into a distinct singlet at lower frequency (410 Hz) '2) thon that for C-2', thus clearly establishing their sequence.**

Fig. 1. Relevant $\tilde{}$ C-NMR resonances of galactosyl-B(1+3)-ribonucleosides <u>3a</u> and 3b, and of model compounds, in D₂O with dioxane (6 = 67.4 ppm) as internal standard (25.16 MHz, Varian XL 100). Data of methyl ribosides are from Gorin and Mazurek¹³⁾, and were measured against external tetramethylsilane. (Abbreviations: $U = uracil$; Hp = hypoxanthine)

Comparing now the ribose carbons of 3g and 3b with those of their parent nucleosides, i.e. uridine and inosine, resp., (ct. Fig. 1), it is clearly apparent that the galactosyl residues are linked to Q-3', since **the appended C-3' signals are displaced downfield by a sizable 7.5 and 7.2 ppm, respectively, as contrasted to an adverse, considerably smaller upfield shift for the C-4'-resonances and only negligible effects** on the other ribose carbons. This is in sweeping accord with the shifts observed for methyl β -D-ribofurano**side and its 3-O-methyl (cf. Fig. 1) or 3-O-isopropyl derivatives '3) , establishing a close parallel of** effects found on Q-alkylation and Q-glycosylation. In contrast, 2'-Q-substitution results in an entirely **different shift pattern, OS exemplified by the last example in Fig. 1.**

Fig. 2. Selectively 1 H-decoupled 13 C-NMR spectrum of $\frac{3a}{2}$ (D₂O, Varian XL 100). The proton decoupling frequencies given are relative to sodium $\mathtt{D}_\mathtt{A}\mathtt{-trimethylsilyl}$ propionate in D_2O .

Closely analogous shift displacements are found for $\beta(1\rightarrow 5)$ - or $\alpha(1\rightarrow 5)$ -linked glycosyl-ribonucleosides as evidenced by the data given in Fig. 3 for <u>2a</u>, <u>Ib</u> and the corresponding model compounds. Here, the **C-5' signal shows a distinct downfield shift as compared to the smaller, yet clearly noticeable adverse effect on C-4' and essentially negligible consequences for the other ribose carbons. Again, displacements** in the same direction and the same order of magnitude are observed for methyl β -D-riboside and its $5'$ -O**methyl derivative (Fig. 3).**

For characterization purposes, glycosyl-ribonucleosides have often been converted into their wellcrystallizing per-O-acetates 3-7) , thus a comparative evaluation of shift displacements between tri-Oacetyl–uridine and the peracetate ot <u>Za</u> was made, with methyl tetra–O-acetyl-β-D-galactopyranoside tor **signal identification. Not unexpectedly, in view of comparing carbons carrying acetoxy and tetraacetylglycosyloxy groups the effects are different, yet nevertheless significant. When going from peracetyl**uridine to 2a-hexacetate downfield shifts are observed for C-5' (63.2-+68.0 ppm) and C-4' (80.0-+81.9), whilst the anomeric C-1', is reversely displaced (87.6+85.8 ppm).

In summary, the shift displacement patterns observed between ribonucleosides and their 0'-glycosylation products are seen to provide an efficient tool with which to assign glycosylation sites. It also seems likely that the approach outlined herein will prove applicable to other ribonucleoside-derived products such as ethers, acetals and esters, most opportune being the ditterentiability between 2' -Q- and 3' -Q-amino **-derivatives.**

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REFERENCES AND NOTES

- 1. Part 44: F.W. Lichtenthaler and A. Moser, *Tetrahedron Lett.* 22 (1981), preceed:
- 2. P.A.J. Gorin, J.F.T. Spencer, and H.J. Phaft, *Can. J. Chem.* 42, 2307 (1964); D.J. Manne and J.R. stark, *Carbohydr. Res. 3,* **102 (1966);** A. Zurwska, F. Villarroya, and F. Petek, *ibid.* 24, 319 (1974).
- **3.** Y. Suzuki and K. Uchida, *Vitamins* (Japan) 44, 196 (1971); *Nippon Nogei Kagaku tiishi 53,* 285 (1979) *[Chem. Abstr.* <u>92</u>, 54044k (1980)]; I. Nogami, Y. Arai, and M. Yoneda (Takeda &em. end.), *Jap. Kokai* **74-117 689 (13 Mar 1973)** *[Chem. Abstr. 8& 1538ood (1975)].*
- **4. Y. Suzuki, K. Uchida, and S. Fujimori,** Nippon Nogei Kagaku Kaishi <u>48</u>, 605 (1974) [Chem. *Abstr. 82, 134774~ (1975)].*
- **5.** Y. Suzuki and K. Uchida, *Nippon Nogei Kuguku Kaishi, 50,* **231 (1976** *[Chem. Abstr. 85, 74157h (1976jl.*
- **6.** Y. Suzuki and K. Uchida, *Nippon Nogei tiguku Kuishi, so, 237 (1976) [Chem. Abstr. S, 74158j (197611.*
- **7.** F.W. Li&tenthaler, Y. Sanernitsu, end T. Nohara, *Angew. Chem. 90,* **819 (1978);** Angew. Chem. Int. *Ed. Engt. 11, 772 (1978).*
- 8. B. Kraska and F.W. Lichtenthaler*, Chem. Ber*. <u>114</u>, 1636 (1981).
- **9. W. Eberhard, F.W. Lichtenthaler, and K.A. Khan,** *Nucleic Acids Res., Spec. Publ*. <u>9</u>,in pres
- 10. Samples of enzymatically prepared $\underline{3a}^{6)}$ and $\underline{3b}^{5)}$ were kindly provided by Prof. Y. Suzuki, Okayama University, Kurashiki.
- 11. In the 'H-NMR spectra of <u>3a</u> (cf. Fig. 1 in ref. 9) and <u>3b</u>, ribose and galactose protons are well separated (except those at the anomeric centre), thus considerably facilitating the selective decoupling.
- **12.** Presupposition for the C-3'/C-2'-differentiation was , of course, theunequivccal,double resonance-based assignment of H-3' (dd of J = 5.1 and 6.1 Hz at δ = 4.39 ppm) and H-2' (dd of $J = 4.1$ and 5.1 Hz at 4.50).
- **13.** P.A.J. Gorin end M. Mazurek, *Carbohydr. Res. 48, 171 (1976).*

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